

Designing P-Chirogenic 1,2-Diphosphinobenzenes at Both P-Centers Using P(III)-Phosphinites

Jérôme Bayardon,* Yoann Rousselin, and Sylvain Jugé*

Institut de Chimie Moléculaire de l'Université de Bourgogne (ICMUB-StéréochIM), UMR CNRS 6302, 9 Avenue Alain Savary, BP 47870-21078 Cedex Dijon, France

Supporting Information

ABSTRACT: A new enantiodivergent synthesis of P-chirogenic 1,2-diphosphinobenzenes (DP*B) bearing the chirality on one or both phosphorus centers is reported using aryne chemistry. The principle is based on successive reactions of 1,2-dibromobenzene with *sec*-phosphide boranes, then DABCO to remove the borane, and finally with chlorophosphines or P(III)-chirogenic phosphinites. The efficiency of this synthesis was demonstrated by the stereoselective preparation of (S,S)-1,2-bis(o-anisylphenylphosphino)benzene. A compari-

son of DIPAMP and homochiral DP*B ligands in asymmetric Rh- or Pd-catalyzed reactions was reported.

hiral phosphines are currently used in coordination chemistry and in asymmetric catalysis by chiral transition metal complexes.1 However, this chemical class also concerns organocatalysis,² molecular materials,³ polymers of coordination, 4 stereoselective reagents, 5 and compounds of therapeutic value. Among the chiral phosphine classes, the C_1 - and C_2 symmetric 1,2-diphosphinobenzenes receive particular attention because the phenylene bridge between the two P atoms gives characteristic steric and electronic properties to the complex formed with the phosphine moieties. Since Burk's pioneering work on the development of the DuPHOS ligands 1 in Rhcatalyzed asymmetric hydrogenation, many diphosphines such as 2-6 with aromatic or heteroaromatic rings as bridges have been developed in asymmetric catalysis. Most of these ligands are chiral by virtue of the carbon backbone, whereas few examples are described with P-chirogenic centers such as 2-5 (Figure 1).9 In the case of diphosphine 5, it was noted that the chirality is provided by carbon and phosphorus centers (Figure 1).

Chiral 1,2-diphosphinobenzene ligands 1, 5, and 6 are usually synthesized according to nonversatile methods which require multistep procedures. In the case of the P-chirogenic diphosphinobenzenes (DP*B) 2, 3, or 4, their syntheses were performed using the *tert*-butylmethylphosphine borane 8 as a building block (Scheme 1a). However, this method is mainly restricted to phosphines carrying a *tert*-butyl and a methyl substituent, which are required for the preparation of 8 by kinetic resolution of *tert*-butyldimethylphosphine borane 7 in the presence of sparteine (Scheme 1a).

Stereoselective synthesis of ligand 12 was reported using the (+)-ephedrine method (Scheme 1b). This synthesis is based on the diastereoselective ring opening of the oxazaphospholidine—borane complex (-)-9 by the organolithium reagent prepared by metal-halide exchange from 2-bromophenyl(diphenyl)-phosphine 10a. Methanolysis of the ring-opened compound 11, followed by reaction with an organolithium reagent (RLi) and

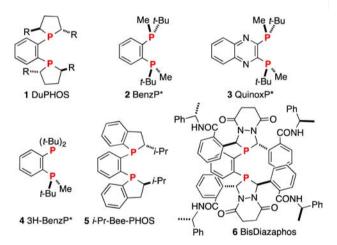


Figure 1. Representative chiral 1,2-diphosphino aromatic-bridged ligands.

 BH_3 removal, led to the P-chirogenic 12 bearing the chirality on one P-center (Scheme 1b). 10

We described an efficient stereoselective synthesis of P-chirogenic o-halogenophenylphosphines 10 from secondary phosphine boranes 13 using aryne chemistry (Scheme 2). Compounds 10 have been applied to the synthesis of o-hydroxyalkyl- and o-boronatophenyl P-chirogenic phosphines. In continuation of this work, we now report the enantiodivergent synthesis of C_1 - and C_2 -symmetric P-chirogenic 1,2-diphosphinobenzenes 12. 11

The principle of the synthesis is based on the phosphination of the o-bromophenylphosphines **10** in the ortho position using

Received: May 3, 2016 Published: May 31, 2016 Organic Letters Letter

Scheme 1. Stereoselective Synthesis of 1,2-Diphosphinobenzene-Type Ligands

Scheme 2. Stereoselective Synthesis of P-Chirogenic DP*B 12

either chlorophosphines 14 or phosphinites 15 (Scheme 2, route A or B). Synthesized P-chirogenic 1,2-diphosphinobenzenes are summarized in Table 1.

o-Bromophenylphosphine boranes 10 were previously prepared in yields up to 75% by reaction of P-chirogenic secphosphine boranes 13 with n-butyllithium (1.2 equiv) and 1,2-dibromobenzene, followed by decomplexation with DABCO, according to the procedure described in the literature. Addition of n-butyllithium to 10 at -78 °C led, by metalhalogen exchange, to the corresponding carbanions in the ortho

position, which react with the chlorophosphines 14 to afford the free P-chirogenic diphosphines 12a-k stereoselectively in 37–70% isolated yields (Scheme 1, route A; Table 1, entries 1–11). On the other hand, when *o*-carbanions derived from 10 reacted with the P(III)-phosphinites 15, diphosphines 12l-q were obtained in 33–67% yields with 99% ee (Scheme 1, route B; Table 1, entries 12–19). Very interestingly, no racemization was detected by reaction with the P(III)-chirogenic phosphinites 15, and this method allowed us to synthesize diphosphines such as 12o-q with two P-chirogenic centers (entries 17–19).

Free P(III)-chirogenic phosphinites 15 were prepared using the (+)-ephedrine method starting from (-)-9. ¹³ Thus, reaction of complex 9 with organolithium reagents R'Li led to the ring-opened products 16, which successively reacted with either HCl in toluene and then sodium phenolate or methanol in acidic conditions to afford P-chirogenic phosphinites 15a-d stereoselectively in 67–74% overall yields after decomplexation with DABCO (Scheme 3). Note that as the stereochemistry of routes

Scheme 3. Synthesis of P(III)-Chirogenic Phosphinites 15

A and B proceeded with retention and inversion, respectively, the phenyl and methyl P(III)-chirogenic phosphinites 15a-c and 15d were obtained with opposite absolute configurations at the P-

Table 1. Structure of P-Chirogenic DP*B 12^{a,b}

entry	\mathbb{R}^1	\mathbb{R}^2	\mathbb{R}^3	diphosphines 12	yield ^c (%)
1	o-An	Ph	Ph	(S)-12a	70
2	o-An	1-Np	1-Np	(S)-12b	44
3	o-An	o-Tol	o-Tol	(S)-12c	37
4	o-An	<i>p</i> -Tol	<i>p</i> -Tol	(S)-12d	52
5	o-An	p-CF ₃ C ₆ H ₄	p-CF ₃ C ₆ H ₄	(S)-12e	58
6	o-An	m-Xyl	m-Xyl	(S)-12f	61
7	o-An	$3,5-(CF_3)_2C_6H_3$	$3,5-(CF_3)_2C_6H_3$	(S)-12g	44
8	o-An	c-Hex	c-Hex	(S)-12h	47
9	o-An	i-Pr	i-Pr	(S)-12i	43
10	Fc	Ph	Ph	(R)-12j	56
11	i-Pr	Ph	Ph	(R)-12k	45
12 ^d	Ph	o-Tol	Ph	(S)-12I	54
13 ^e	Ph	$3.5-(t-Bu)_2C_6H_3$	Ph	(S)-12m	45
14 ^f	Ph	o-An	Ph	(S)-12a	67
15 ^g	Ph	Ph	o-An	(R)-12a	55
16 ^h	Ph	o-biPh	Ph	(S)-12n	33
17 ^h	o-An	o-biPh	Ph	(S,S)-120	35
18 ^f	o-An	o-An	Ph	(S,S)-12 p	52
19 ^f	i-Pr	o-An	Ph	(R,S)-12q	56

^aDiphosphines **12a–k** (R² = R³ = R') were synthesized according to route A (entries 1–11); diphosphines **12a,l–q** were synthesized according to route B (entries 12–19). ^bDiphosphines have 99% ee, determined by HPLC on a chiral column (except (R)-**12k**, 98% ee, entry 11). ^cIsolated yields. ^dFrom **15b**. ^eFrom **15c**. ^fFrom **15a**. ^gFrom **15d**. ^hFrom **15e**.

Organic Letters Letter

center, which allowed us to use them in enantiodivergent synthesis (Scheme 3). ¹⁴ In addition, methyl (R)-(o-biphenyl)-phenylphosphinite **15e** was also prepared as described above but starting from complex (+)-**9** derived from (-)-ephedrine (see Supporting Information).

Both enantiomers of the P-chirogenic 1,2-diphosphinobenzene 12a were prepared according to route B by reaction of the carbanions derived from 2-bromophenyl(diphenyl)phosphine 10a, with phenyl or methyl o-anisylphenylphosphinites 15a (or 15d), respectively (Scheme 2; Table 1, entries 14 and 15). On the other hand, diphosphinobenzene (S)-12a was also obtained using (S)-o-anisyl(bromophenyl)phenyl phosphine 10b and chlorodiphenylphosphine 14a (R = Ph) (Scheme 2, route A; Table 1, entry 1).

All P-chirogenic 1,2-diphosphinobenzenes (DP*B) 12 prepared by methods A and B were analyzed by HPLC on a chiral column (>98% ee) and were fully characterized (see Supporting Information).

Crystallization of DP*B (S)-12a and (R)-12j provided crystals suitable for X-ray diffraction analyses, and their OLEX2 views are shown in Supporting Information. Interestingly, this method offers the possibility to synthesize C_1 - and C_2 -symmetric P-chirogenic 1,2-diphosphinobenzenes bearing the chirality on both P-centers. Thus, DP*B 12o-q were synthesized in yields up to 56% by successive reactions of the o-bromophenyl phosphines (S)-10b (or (R)-10d) with n-BuLi to perform the metal-halide exchange and then with the P-chirogenic phosphinites 15e (or 15a), respectively (Scheme 2, route B). Note that the reactions leading to C_2 -symmetric DP*B were diastereoselective because no epimers (or meso) could be detected. This demonstrated that the substitution of the P(III)-chirogenic phosphinites 15a or 15e was highly stereospecific.

Compound DP*B **12p** was recrystallized in a dichloromethane/methanol mixture, providing crystals suitable for X-ray diffraction analyses. The OLEX2 view is shown in Figure 2.

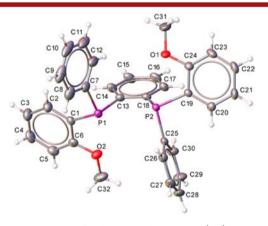


Figure 2. OLEX2 view of P-chirogenic diphosphine (S,S)-12p. Thermal ellipsoids are drawn at the 50% probability level. Selected bond lengths (Å), angles (deg), and dihedral angles (deg): C13-P1 = 1.838(4), C18-P2 = 1.835(3); C7-P1-C1 = 102.21(13), C25-P2-C19 = 100.77(14); C14-C13-P1-C1 = 17.1(3)°, C17-C18-P2-C19 = 14.9(3)°.

Compound 12p crystallizes in the noncentrosymmetric C_2 space group, and the S absolute configuration at the P atoms is supported by refinement of the Flack parameter. The structure shows that the o-anisylphenylphosphino moieties do not significantly alter the planarity of the P-o-An bonds with the phenylene bridge, as proven by the dihedral angles C14–C13–P1–C1 and C17–C18–P2–C19 of 17.1(3) and $14.9(3)^\circ$,

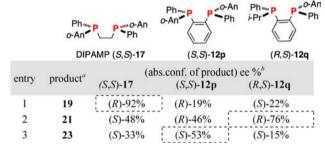
respectively. The aromatic substituents of the phosphine moieties are perpendicular to each other, with mean angles of 95.39(12)° (see Supporting Information).

Homochiral 1,2-bis(*o*-anisylphenylphosphino)benzene (*S*,*S*)-12**p** and DIPAMP (*S*,*S*)-17 ligands were used in asymmetric Rhand Pd-catalyzed reactions, and their efficiencies were compared (Scheme 4). Modified DP*B ligand (*R*,*S*)-12**q**, obtained by substituting one *o*-anisyl substituent of 12**p** for *i*-propyl, was also tested in these conditions.

Scheme 4. Asymmetric Transition-Metal-Catalyzed Reactions Studied Using P-Chirogenic Ligands 17, 12p, and 12q

Cationic rhodium complexes $[Rh(COD)L]BF_4$ were previously prepared in yields up to 70% by mixing $[Rh(COD)_2]BF_4$ with the ligands L= 17, 12p, and 12q. Whereas the phenylalanine derivative (R)-19 was obtained with 92% ee using DIPAMP (S,S)-17, 19 and 22% ee values were obtained with homochiral (S,S)-12p and modified DP*B 12q ligands, respectively (Table 2, S)-12p and (Table

Table 2. Comparative Asymmetric Catalysis with P-Chirogenic Ligands 17 (DIPAMP), 12p, and 12q



^aProducts 19, 21, and 23 were obtained in almost quantitative yields according to the conditions mentioned on Scheme 4. ^bThe ee was determined by HPLC on a chiral column.

entry 1). On the other hand, when dimethyl itaconate **20** was hydrogenated in dichloromethane with the Rh complexes prepared as above, the configuration (S or R) of the resulting diester **21** surprisingly depended on the nature of the DIPAMP (S,S)-**17**, DP*B (S,S)-**12p**, and **12q** (entry 2). In these conditions, the best result was obtained with DP*B **12q**, which led to the methyl (R)-methylsuccinate **21** with 76% ee (entry 2). Finally, when the P-chirogenic ligands were tested in Pd-catalyzed allylation of methyl malonate, allylated product (S)-**23** was obtained with up to 53% ee (entry 3). In this case, asymmetric induction was slightly higher using DP*B (S,S)-**12p** than using homochiral DIPAMP (S,S)-**17** ligand.

In conclusion, a new and efficient stereoselective synthesis of P-chirogenic diphosphinobenzene ligands was reported. This Organic Letters Letter

method allowed us to synthesize C_1 - or C_2 -symmetric DP*B ligands bearing the chirality on one or both phosphorus centers, by phosphination of 2-bromophenylphosphines, previously prepared due to the aryne chemistry, with chlorophosphines or P(III)-chirogenic phosphinites. Interestingly, according to the use of either phenyl or methyl P(III)-chirogenic phosphinite, prepared from the same aminophosphine borane precursor, both DP*B enantiomers were independently obtained. A brief comparison of the homochiral DP*B and DIPAMP ligands in asymmetric metal-based catalysis has shown that the former gave better results in Pd-catalyzed allylation, while DIPAMP affords the best asymmetric induction in the hydrogenation of methyl α acetamido cinnamate. This new enantiodivergent synthesis, based in part on the use of P(III)-chirogenic phosphinite, is a powerful method for the design and development of the DP*B ligands bearing one or two P-chirogenic centers.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b01275.

Experimental data, selected spectral data for all new compounds, and X-ray data for (S)-12a, (R)-12j, and (S,S)-12p (PDF)

Crystallographic data (CIF)

AUTHOR INFORMATION

Corresponding Authors

*E-mail: jerome.bayardon@u-bourgogne.fr.

*E-mail: sylvain.juge@u-bourgogne.fr.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The authors are grateful for the financial support provided by the CNRS (Centre National de la Recherche Scientifique), the Ministère de l'Education Nationale et de la Recherche, the Conseil Régional de Bourgogne (Grants 3MIM, Pari II-CDEA/smt8), and the Agence Nationale pour la Recherche (Grant 07BLAN292-01 *MetChirPhos*). The authors also thank M.J. Penouilh, M. Picquet, and F. Picquet at the Welience/Pôle Chimie Moléculaire for the NMR and mass spectrometry analyses, and M. J. Eymin (ICMUB-StéréochIM) for her skilled technical assistance.

REFERENCES

- (1) (a) Tang, W.; Zhang, X. Chem. Rev. 2003, 103, 3029. (b) Phosphorus Ligands in Asymmetric Catalysis; Börner, A., Ed.; Wiley-VCH: Weinheim, Germany, 2008; Vols. 1–3. (c) Zhou, Q.-L., Ed.; Priviliged Chiral Ligands; Wiley-VCH: Weinheim, Germany, 2011. (d) Homogeneous Catalysis: Design and Synthesis; Kamer, P. C. J., van Leeuwen, P. W. N. M., Eds.; Wiley: Chichester, U.K., 2012.
- (2) (a) Denmark, S. E.; Beutner, G. L. Angew. Chem., Int. Ed. 2008, 47, 1560.
 (b) Werner, T. Adv. Synth. Catal. 2009, 351, 1469–1481.
 (c) Enders, D.; Nguyen, T. V. Org. Biomol. Chem. 2012, 10, 5327.
 (d) Rémond, E.; Jugé, S. Chem. Today 2014, 32, 49.
- (3) (a) Oosterom, G. E.; Reek, J. N. H.; Kamer, P. C. J.; van Leeuwen, P. W. N. M. Angew. Chem., Int. Ed. 2001, 40, 1828. (b) Rodriguez, L.-I.; Rossell, O.; Seco, M.; Orejon, A.; Masdeu-Bulto, A. M. J. Supercrit. Fluids 2011, 55, 1023. (c) He, Y.-M.; Feng, Y.; Fan, Q.-H. Acc. Chem. Res. 2014, 47, 2894. (d) Caminade, A. M.; Ouali, A.; Laurent, R.; Turrin, C.-O.; Majoral, J. P. Chem. Soc. Rev. 2015, 44, 3890.

- (4) (a) Salomon, C.; Fortin, D.; Khiri, N.; Jugé, S.; Harvey, P. D. Eur. J. Inorg. Chem. 2011, 2597. (b) Morisaki, Y.; Suzuki, K.; Imoto, H.; Chujo, Y. Macromol. Chem. Phys. 2011, 212, 2603. (c) Lapprand, A.; Dutartre, M.; Khiri, N.; Levert, E.; Fortin, D.; Rousselin, Y.; Soldera, A.; Jugé, S.; Harvey, P. D. Inorg. Chem. 2013, 52, 7958.
- (5) (a) Trost, B. M.; Curran, D. P. Tetrahedron Lett. 1981, 22, 4929.
 (b) Headley, C. E.; Marsden, S. P. J. Org. Chem. 2007, 72, 7185.
- (6) (a) Wang, Y.; Liu, M.; Cao, R.; Zhang, W.; Yin, M.; Xiao, X.; Liu, Q.; Huang, N. J. Med. Chem. 2013, 56, 1455. (b) Hayashi, R.; Nakatsui, K.; Sugiyama, D.; Kitajima, T.; Oohara, N.; Sugiya, M.; Osada, S.; Kodama, H. J. Inorg. Biochem. 2014, 137, 109.
- (7) Burk, M. J. J. Am. Chem. Soc. 1991, 113, 8518.
- (8) (a) Jiang, Z.; Sen, A. J. Am. Chem. Soc. 1995, 117, 4455. (b) Barnhart, R. W.; McMorran, D. A.; Bosnich, B. Chem. Commun. 1997, 589. (c) Holz, J.; Quirmbach, M.; Schmidt, U.; Heller, D.; Stürmer, R.; Börner, A. J. Org. Chem. 1998, 63, 8031. (d) Gilbertson, S. R.; Hoge, G. S.; Genov, D. G. J. Org. Chem. 1998, 63, 10077. (e) Jiang, Q.; Jiang, Y.; Xiao, D.; Cao, P.; Zhang, X. Angew. Chem., Int. Ed. 1998, 37, 1100. (f) Murakami, M.; Itami, K.; Ito, Y. J. Am. Chem. Soc. 1999, 121, 4130. (g) Marinetti, A.; Genêt, J.-P.; Jus, S.; Blanc, D.; Ratovelomanana Vidal, V. Chem. - Eur. J. 1999, 5, 1160. (h) Xiao, D.; Zhang, Z.; Zhang, X. Org. Lett. 1999, 1, 1679. (i) Li, W.; Zhang, Z.; Xiao, D.; Zhang, X. J. Org. Chem. 2000, 65, 3489. (j) RajanBabu, T. V.; Yan, Y.-Y.; Shin, S. J. Am. Chem. Soc. 2001, 123, 10207. (k) Kottsieper, K. W.; Kühner, U.; Stelzer, O. Tetrahedron: Asymmetry 2001, 12, 1159. (1) Matsumura, K.; Shimizu, H.; Saito, T.; Kumobayashi, H. Adv. Synth. Catal. 2003, 345, 180. (m) Malaisé, G.; Ramdeehul, S.; Osborn, J. A.; Barloy, L.; Kyritsakas, N.; Graff, R. Eur. J. Inorg. Chem. 2004, 3987. (n) Wang, Y.-Q.; Lu, S.-M.; Zhou, Y.-G. Org. Lett. 2005, 7, 3235. (o) Clark, T. P.; Landis, C. R.; Freed, S. L.; Klosin, J.; Abboud, K. A. J. Am. Chem. Soc. 2005, 127, 5040. (p) Axtell, A. T.; Cobley, C. J.; Klosin, J.; Whiteker, G. T.; Zanotti-Gerosa, A.; Abboud, K. A. Angew. Chem., Int. Ed. 2005, 44, 5834. (q) Gini, F.; Hessen, B.; Minnaard, A. J. Org. Lett. 2005, 7, 5309. (r) Mondal, M.; Ibrahim, A. A.; Wheeler, K. A.; Kerrigan, N. J. Org. Lett. 2010, 12, 1664. (s) McDonald, R. I.; Wong, G. W.; Neupane, R. P.; Stahl, S. S.; Landis, C. R. J. Am. Chem. Soc. 2010, 132, 14027. (t) Chen, S.; Salo, E. C.; Wheeler, K. A.; Kerrigan, N. J. Org. Lett. 2012, 14, 1784. (u) Chang, M.; Liu, S.; Huang, K.; Zhang, X. Org. Lett. 2013, 15, 4354.
- (9) (a) Miura, T.; Imamoto, T. Tetrahedron Lett. 1999, 40, 4833. (b) Shimizu, H.; Saito, T.; Kumobayashi, H. Adv. Synth. Catal. 2003, 345, 185. (c) Yamamoto, Y.; Koizumi, T.; Katagiri, K.; Furuya, Y.; Danjo, H.; Imamoto, T.; Yamaguchi, K. Org. Lett. 2006, 8, 6103. (d) Tamura, K.; Sugiya, M.; Yoshida, K.; Yanagisawa, A.; Imamoto, T. Org. Lett. 2010, 12, 4400. (e) Imamoto, T.; Tamura, K.; Zhang, Z.; Horiuchi, Y.; Sugiya, M.; Yoshida, K.; Yanagisawa, A.; Gridnev, I. D. J. Am. Chem. Soc. 2012, 134, 1754. (f) Zhang, Z.; Tamura, K.; Mayama, D.; Sugiya, M.; Imamoto, T. J. Org. Chem. 2012, 77, 4184. (g) Reynolds, S. C.; Hughes, R. P.; Glueck, D. S.; Rheingold, A. L. Org. Lett. 2012, 14, 4238.
- (10) Rast, S.; Stephan, M.; Mohar, B. Eur. J. Org. Chem. 2015, 2214.
- (11) (a) Jugé, S.; Bayardon, J.; Lauréano, H.; Henry, J. C.; Colobert, F.; Leroux, F.; Rémond, E. P-Chirogenic Organo Phosphorus Compounds. Int. Patent WO 2013/007724 A1, August 29, 2011. (b) Bayardon, J.; Lauréano, H.; Diemer, V.; Dutartre, M.; Das, U.; Rousselin, Y.; Henry, J. C.; Colobert, F.; Leroux, F. R.; Jugé, S. J. Org. Chem. 2012, 77, 5759.
- (12) (a) Rémond, E.; Bayardon, J.; Takizawa, S.; Rousselin, Y.; Sasai, H.; Jugé, S. *Org. Lett.* **2013**, *15*, 1870. (b) Bayardon, J.; Bernard, J.; Rémond, E.; Rousselin, Y.; Malacea-Kabbara, R.; Jugé, S. *Org. Lett.* **2015**, *17*, 1216.
- (13) (a) Moulin, D.; Bago, S.; Bauduin, C.; Darcel, C.; Juge, S. *Tetrahedron: Asymmetry* **2000**, *11*, 3939. (b) Bauduin, C.; Moulin, D.; Kaloun, E. B.; Darcel, C.; Jugé, S. *J. Org. Chem.* **2003**, *68*, 4293.
- (14) It should be noted that little or no reaction was obtained by reaction with the borane complex of the phosphinite 15.